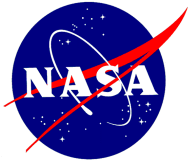


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NANOTECHNOLOGY AND GELLED CRYOGENIC FUELS

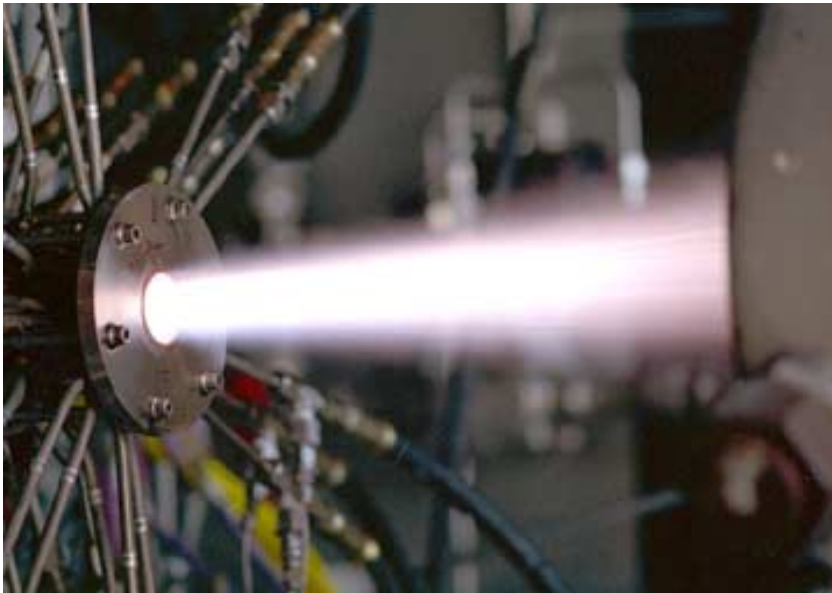
**Presentation to
Dr. Minoo Dastoor
NASA Nano / Bio Initiative**

**Bryan Palaszewski
NASA John H. Glenn Research Center, Lewis Field
Cleveland, OH, 44135**

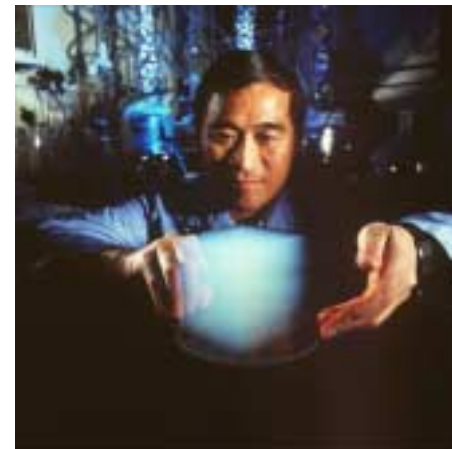


Nanoparticulates for Gelled and Metallized Gelled Propellants

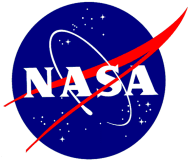
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Aerogel is a highly efficient insulator (CNN)



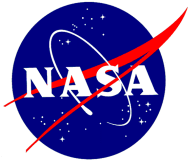
**O₂ / RP-1 / Aluminum combustion –
Aerogel and nanoparticulate metals
can gel the fuel, making it denser, more
energetic, and safer**



The Benefits of Nanogellant Gelled Cryogenic Propellants and Nanoparticulates

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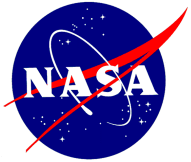
- *Nanogellant Gelled Propellants*
 - Increased safety
 - Increased fuel density
 - Reduced leakage
 - Reduced slosh
 - Reduced cryogenic boiloff
 - Potential reduction in specific fuel consumption
 - Potential increases in engine thrust
- *Nanoparticulate Metallized Gelled Propellants*
 - All of the above and
 - Large increases in fuel density
 - Larger potential reductions in specific fuel consumption
 - Larger potential increases in engine thrust



Nanogellant and Nanoparticulates

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- *What are nanogellant and nanoparticulate fuels?*
 - Nanogellants are gellants that have a nanometer scale structure, which have enormously high surface area per gram
 - Gelled fuel reduce leakage and increase safety
 - Nanoparticulates are metal particles that are 20 nanometers in diameter, much smaller than traditional 7 micron particles used for metal additives
 - Smaller particles allow for more efficient combustion and lower specific fuel consumption



Nanogellants and Nanoparticulates

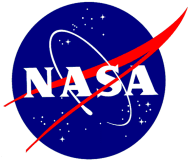
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Why are Gelled Cryogenic Fuels revolutionary?

- Gelled cryogenic fuels reduce leakage and increase safety
- Gelled cryogenic fuels are critical for increasing operability of cryogens for aerospace vehicles
- Nanogellant for gelled cryogens has a surface area of nearly $1000 \text{ m}^2/\text{g}$, leading to cryogenic fuels gelled with 1-7 weight % gellant, 25 to 50% less mass than traditional gellant material

• *Synergy*

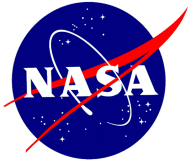
- Gelled and metallized gelled propellants have been an area of considerable interest in the rocket propulsion and explosives



Propellant Technologies: Teams

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- *National team*
 - NASA MSFC (large scale rocket engine testing)
 - NASA GRC (gelled and metallized gelled propellants, small scale engine tests)
- *NASA partners and contacts*
 - U.S. Army Picatinny Arsenal (potential collaboration in nanoparticles)
 - USAF Research Laboratory (hydrocarbons)
 - U.S. Army Aviation and Missile Command (metallized gelled propellants)
 - U.S. Naval Surface Warfare Center, Indian Head (nanoparticle aluminum, explosives)
 - Technanogy (nanometer aluminum particles)
 - Small Business Innovation Research (Argonide, Orbitec, etc.)
 - Many other industry, Government, and university partners

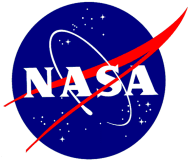


Metallized Gelled Propulsion

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| Oxidizer | Fuel | Metal |
|----------|-------------|-------|
| O_2 | H_2 | Al |
| O_2 | Hydrocarbon | Al |
| NTO | MMH | Al |

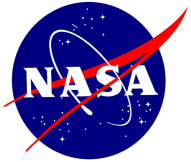
- Metal additives are suspended in gelled fuel and they undergo combustion with oxidizer



Metallized Gelled Propellants: Increasing Rocket Specific Impulse for Mars Missions

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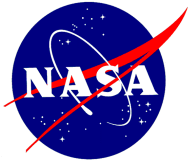
- **Mars missions using metallized gelled O₂ /H₂ /Al have an Isp of 475 to 480 seconds.**
- **O₂/H₂ without gellants or metal particles has an Isp of 470 seconds**
- **Rocket specific impulse (Isp) increased for several reasons:**
 - **Adding metal shifts O/F ratio from 6.0 to 1.6 (with 60 wt% Al), reducing the molecular weight of the rocket exhaust**
 - **Reducing the molecular weight increases engine Isp**
 - **Adding metal actually decreases the combustion temperature by 500 K**
 - **The added metal weight percent (wt%) is 60 to 70 % of the total H₂/Aluminum fuel mass**
 - **Adding 60 wt% Aluminum increases engine Isp by 5 seconds**
 - **Adding 70 wt% Aluminum increases engine Isp by 10 seconds**
 - **O/F change increases Mars vehicle volume by only 1.1 % over the O₂/H₂ case**
 - **References:**
 - **Palaszewski, B., "Metallized Propellants for the Human Exploration of Mars ," NASA-Lewis Research Center, NASA TP-3062, presented at the Case For Mars IV Conference, Boulder, CO, June 4-8 1990. Also in the AIAA Journal of Propulsion and Power, Vol. 8, No. 6, Nov.-Dec. 1992, pp. 1192-1199.**
 - **Palaszewski, B. and Rapp, D., "Design Issues for Propulsion Systems Using Metallized Propellants," NASA-Lewis Research Center, AIAA 91-3484, NASA TM-105190, presented at the AIAA/NASA/OAI Conference On Advanced SEI Technologies, Cleveland, OH , September 4-6, 1991.**



Metallized Gelled Propellants: How Gellants Work

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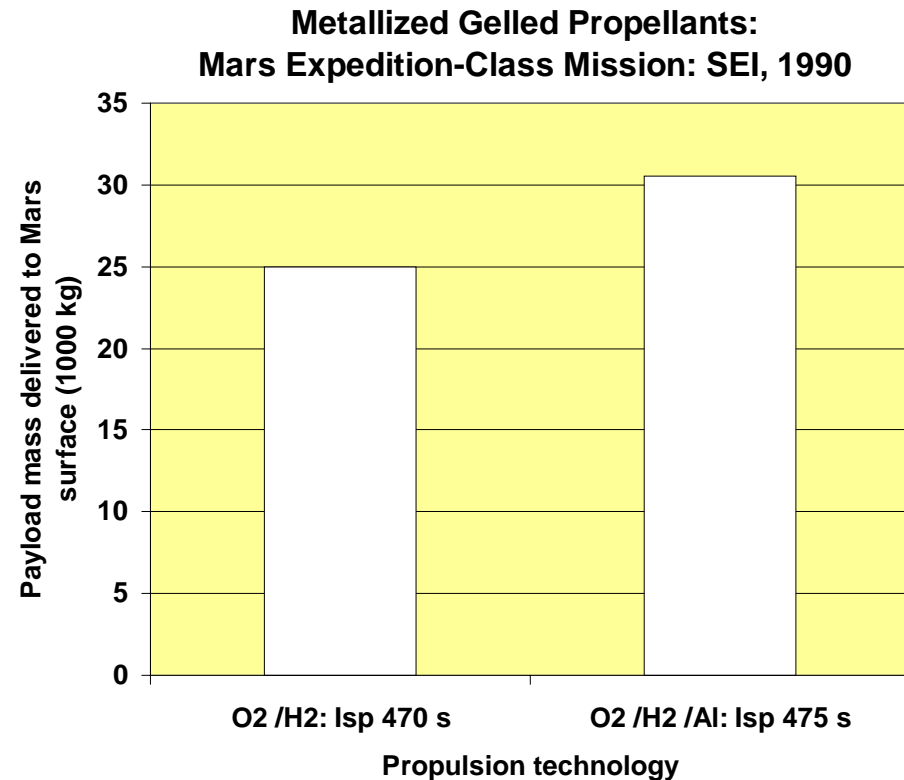
- **Gellants create a cross-linked structure in the liquid fuel, much like a long chain polymer**
- **The gelled liquid fuel is gelled with a small amount of gellant**
 - **RP-1: 0.9 wt% nanogellant**
 - **Liquid hydrogen: 7 to 8 wt% nanogellant**
- **The resulting gelled liquid is thixotropic (shear thinning), and its viscosity is 5 to 10 times that of the liquid alone**
- **The viscosity drops to the liquid viscosity when the fuel flows**
- **Metal particles, if small enough (nanometer sized), can act as a self gellant, reducing or eliminating the need for a separate gellant**

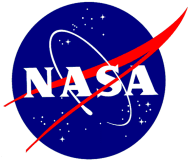


Mars Missions: Space Exploration Initiative (SEI) and Metallized Gelled Propellants

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- Mars missions using
 - O₂ /H₂
 - Metallized gelled propellants: O₂ /H₂ /Al
- 20 to 33% higher payload to Mars surface for each flight



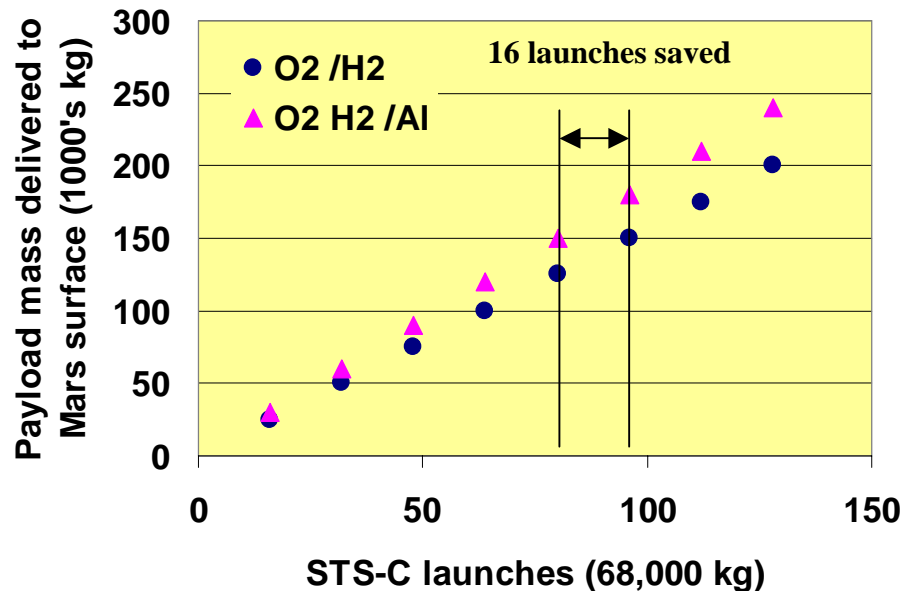


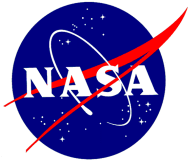
Mars Missions: Space Exploration Initiative (SEI) and Metallized Gelled Propellants

NASA John H. Glenn Research Center, Lewis Field
Turbomachinery and Propulsion Systems Division

- Mars missions using
 - O₂ /H₂
 - Metallized gelled propellants: O₂ /H₂ /Al
- 20% more payload to Mars surface for each flight, with metallized gelled H₂ /Aluminum
- Significant launch vehicle savings with O₂/H₂/Al propellants
- 16 STS-C launches saved over 5 Mars missions
- Faster payload delivery schedule, and billions saved

Metallized Gelled Propellants:
Mars Evolution-Class Missions: SEI, 1990

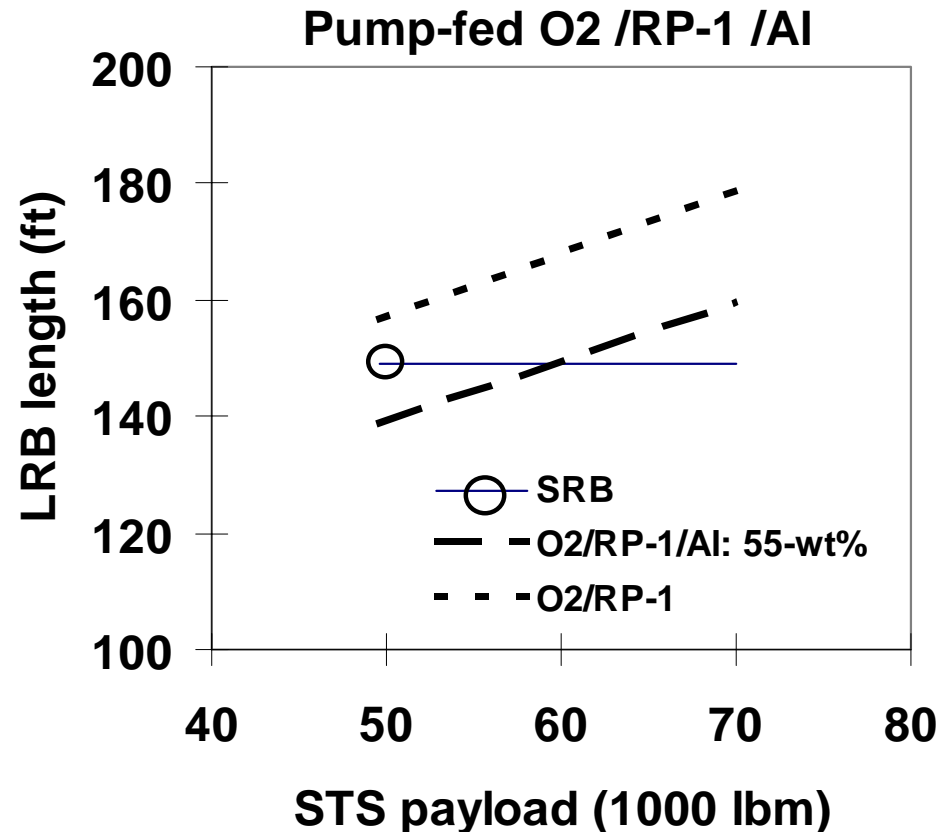


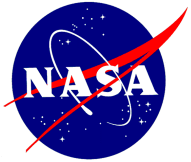


O₂/RP-1/Aluminum Liquid Rocket Booster for Space Shuttle (Future STS)

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- **Payload increases of 14% possible with 55-wt% RP-1/Al (56,600 lbm)**
- **Small 1-ft diameter increase lifts payload to 70,000 lbm**
- **O₂/RP-1: 324 s Isp
O/F = 2.7**
- **O₂/RP-1/Al: 317 s Isp
O/F = 1.1**

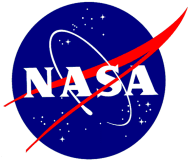




Nanogellant Gelled Propellants: Past Work

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Turbomachinery and Propulsion Systems Division

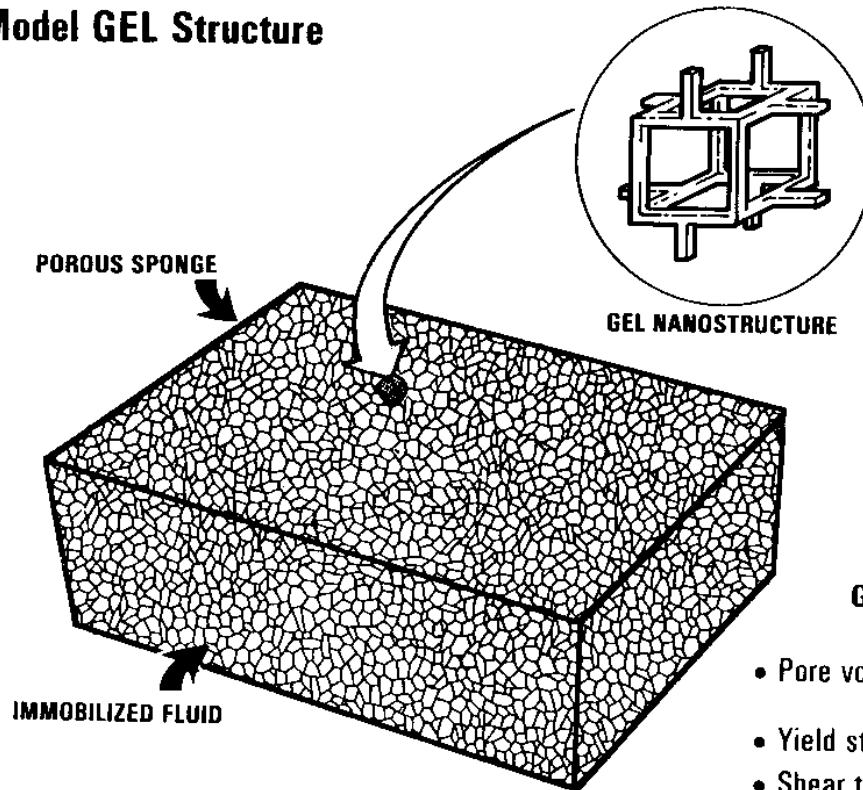
- **Worked with TRW (1989-1996) developing nanoparticulate gellants**
- **Nanogellants are hydrocarbon alkoxide materials, created with a supercritical processing method**
 - **Nanogellant for gelled cryogens has a surface area of nearly 1000 m² /g, leading to cryogenic fuels gelled with 0.9 to 7 to 8 weight % gellant,**
 - **25 to 50% less mass than traditional gellant material**
- **Liquid hexane, RP-1, propane (cryogenic), etc. gelled with less than 0.9 wt% of nanogellant**
- **Liquid hydrogen gelled with 7 to 8 weight % nanogellant (NAS3-25793, 1994 and NAS3-26714, 1996)**
- **Extensive data base on gelled propellants at NASA Glenn**
- **Joint NASA /TRW work in nanogellants being reinvigorated**



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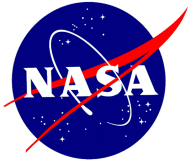
Model GEL Structure

TRW



Gel Merit Figures

- Pore volume = $\frac{V_{\text{GEL}} - V_{\text{SKELETON}}}{W_{\text{SKELETON}}}$
- Yield strength
- Shear thinning
- Energy density



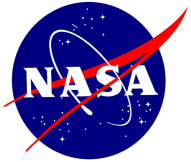
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Transmission Electron Microscopy (TEM) photo of nanogellant

Gellant Particle Morphology

TRW





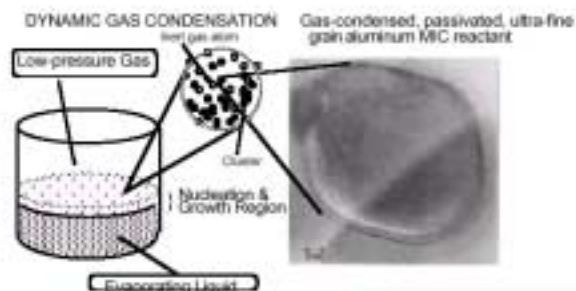
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TRW aerogel type nanogellant, Circa 1990





Dynamic Gas Condensation Is Used At LANL To Fabricate MIC Reactants



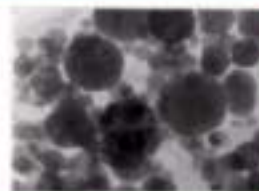
- Ultra fine particles (~20 nm)
- Uniform size distribution
- Size easily controlled
- Amenable to continuous production (currently ~25g/hr)
- Process self-purifies source metal



LANL Al Powder



Russian Al Powder ("ALEX")



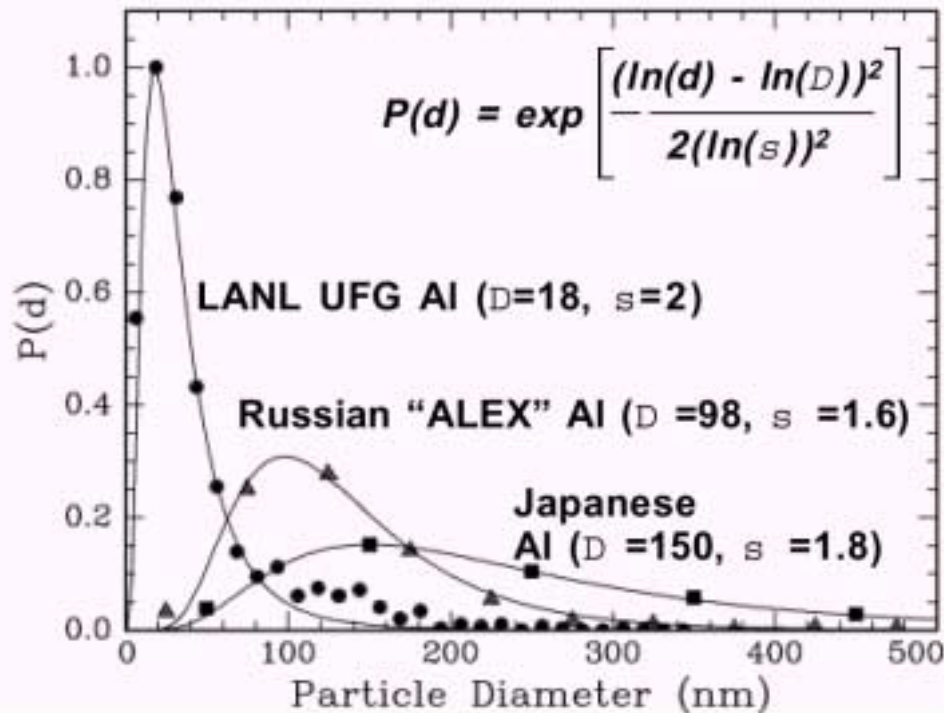
Japanese Al Powder

other fabrication techniques used

500 nm



LANL Al Powders Have A Smaller Mean Size And A Narrower Size Distribution Than Commercial Powders

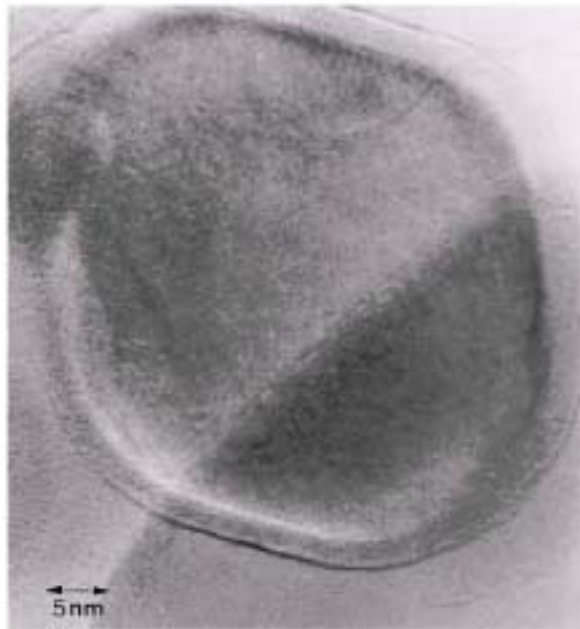


- log-normal distribution function fits measured data well
- D = peak position
- s fixes distribution width

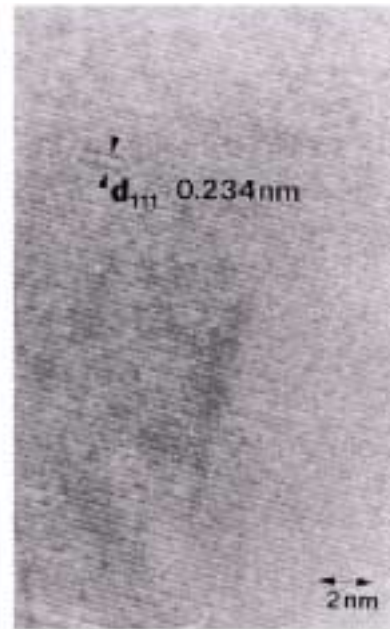


LANL UFG ALUMINUM PARTICLES ARE SINGLE CRYSTALS WITH NEGLIGIBLE STRUCTURAL DEFECT DENSITY

GAS-CONDENSED Al PARTICLE



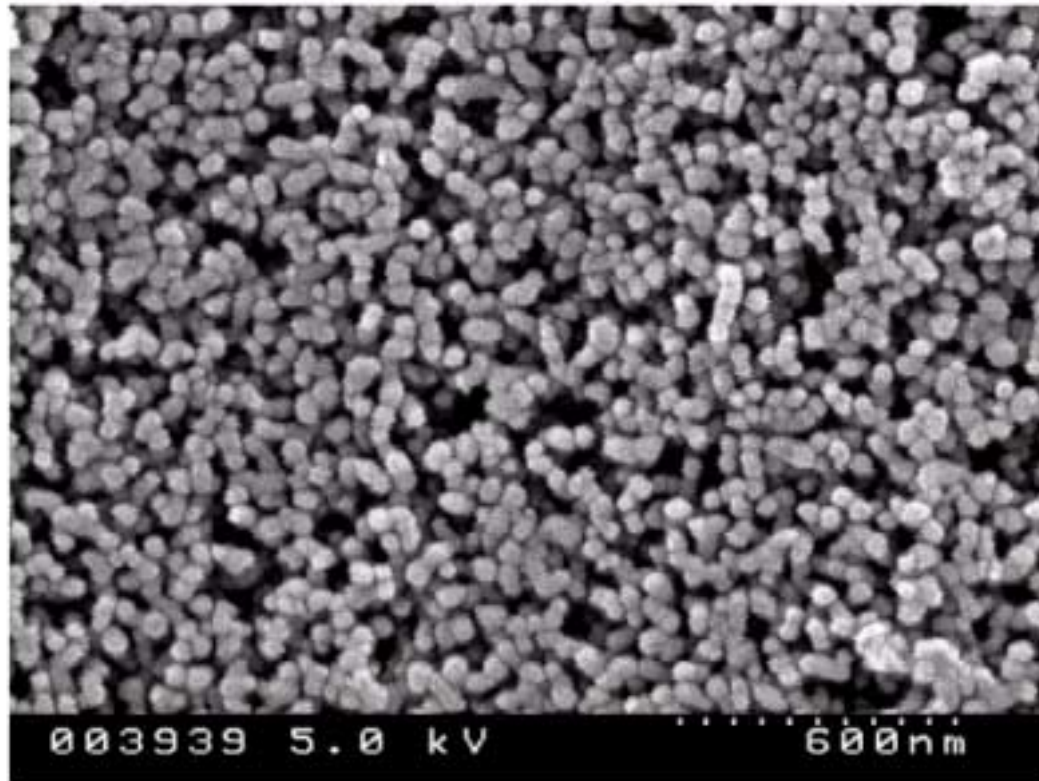
Al (111) LATTICE FRINGES



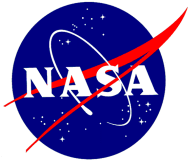
- 1 interior aluminum is crystalline
- 1 no structural defects apparent
- 1 2.5 nm thick Al_2O_3 passivation layer



SEM of LANL UFG ALUMINUM PARTICLES



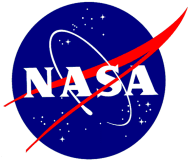
Materials Science and Technology



Nanoparticle and Nanogellant Fuels: Small Business Innovation Research (SBIR)

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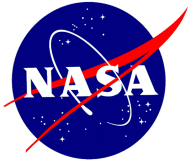
- *Argonide (Sanford, FL)*
 - **Title: Metastable electroexploded nanophase aluminum based gels as a component of propulsion fuels**
 - **SBIR Phase II completed - 2001**
 - **Alex and other metals produced by the electroexplosion of metal wire are metastable, producing additional energy and burning rate**
 - **Discovered reduction of ignition delay with gelled Oxygen /RP-1/ Aluminum fuels**
- *Orbitec (Madison, WI)*
 - **Title: Gelled LH2 /UFAL /LOX propellant system**
 - **SBIR Phase I underway -2001**
 - **Uses ultra-fine aluminum powder (UFAL) to develop a gelled LH2 fuel and LOX propellant system.**
 - **This innovation will increase the performance, density, and combustion efficiency of LH2/Al/LOX for use in rockets and combined-cycle vehicles**



Propellant Technologies – Applications and Fuels

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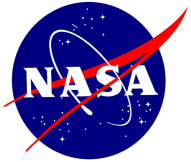
- *Nanogelled Cryogenic Propellants with Nanoparticulate Additives*
 - **Revolutionary Aeropropulsion vehicles**
 - **Next generation aerospace vehicles**
 - **Many others**
- *Planned fuels:*
 - **Liquid methane**
 - **Liquid propane**
 - **Liquid nitrogen**
 - **RP-1**
 - **Jet A**
 - **JP-8**
 - **Liquid hydrogen (last to be addressed in testing)**



Approach

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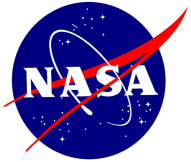
- *First year goals:*
 - **Determine the location for Nanogellant Production**
 - **Produce nanogellant**
 - **Characterize the nanogellant production uniformity**
 - **Determine the effects of storage (shelf life) of the nanogellant**
 - **Determine the location for Nanoparticulate Production**
 - **Produce nanoparticles**
 - **Characterize the particle size and uniformity**
 - **Determine the location for the multi-fuel test area**
 - **Ambient temperature check out**
 - **Produce gelled aviation fuel (high H/C ratio fuel)**
 - **Produce an aviation fuel doped with nanoparticulates**
 - **Produce a gelled aviation fuel doped with nanoparticulates**
 - **Determine the characteristics of the gelled-doped aviation fuel**



Approach

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Turbomachinery and Propulsion Systems Division

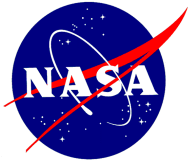
- *Second year and third year goals*
 - **Liquid nitrogen (LN2) check out**
 - **Produce gelled LN2 using nanogellants**
 - **Produce a mixture of LN2 and nanoparticles**
 - **Produce doped-gelled LN2**
 - **Determine characteristics of the doped-gelled LN2**
 - **Determine cryogenic fuels of interest (potential fuel include but are not limited to liquid propane, liquid methane, and liquid hydrogen)**
 - **Fuel 1 to N testing**
 - **Produce gelled Fuel 1 to N using nanogellants**
 - **Produce a mixture of Fuel 1 to N and nanoparticles**
 - **Produce doped-gelled Fuel 1 to N**
 - **Determine characteristics of the doped-gelled Fuel 1 to N**
 - **Optimize production process if necessary to obtain desired results**



Approach

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- *Fourth and fifth year goals*
 - **Build up combustion test area**
 - **Produce sufficient quantities of gelled-doped fuel in production area to support combustion tests**
 - **Develop diagnostic techniques**
 - **Measure uniformity of nanogellant/nanoparticulate dispersion in fuel**
 - **Effect of nanoparticulates on rotating machinery**
 - **Combustion process**
 - **Emissions**
 - **Perform initial combustion tests**
 - **Optimize fuel formulation and repeat combustion tests**



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Gelled and Metallized Gelled Propellants

Contact:

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Cleveland, OH 44135

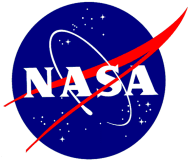
(216) 977-7493 Voice

(216) 433-5802 FAX

bryan.a.palaszewski@grc.nasa.gov

Fuels and Space Propellants Web Site

<http://www.grc.nasa.gov/WWW/TU/launch/foctopsb.htm>

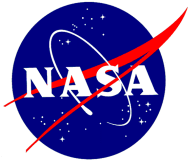


Gelled Hydrogen Propellants

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Past Results

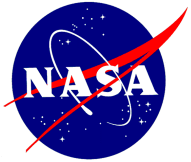
- **First gelled cryogen (nitrogen) was formulated circa 1960**
- **Evaporation (boiloff) of gelled hydrogen**
 - **reduced by factor of 2 to 3 (NAS3-4186, 1966)**
 - **reduced by factor 25 to 50 % (NAS3-2568, 1964)**
 - **variations due to tank geometries, heat leaks**
 - **both used silica gellants, at high weight percentages (36 weight %)**
- **Work with Lockheed (LMSC) and MSFC, with frozen ethane (NAS8-20342, 1968)**
- **Later work used frozen ethane or methane gellant, at 4 to 10 weight % (Aerojet, SNP-1, 1970)**
- **Work with TRW using nanoparticulate gellants, at 7 to 8 weight % (NAS3-25793, 1994 and NAS3-26714, 1996)**
- **Extensive data base on gelled propellants at NASA Lewis**



Metallized Gelled Propellants

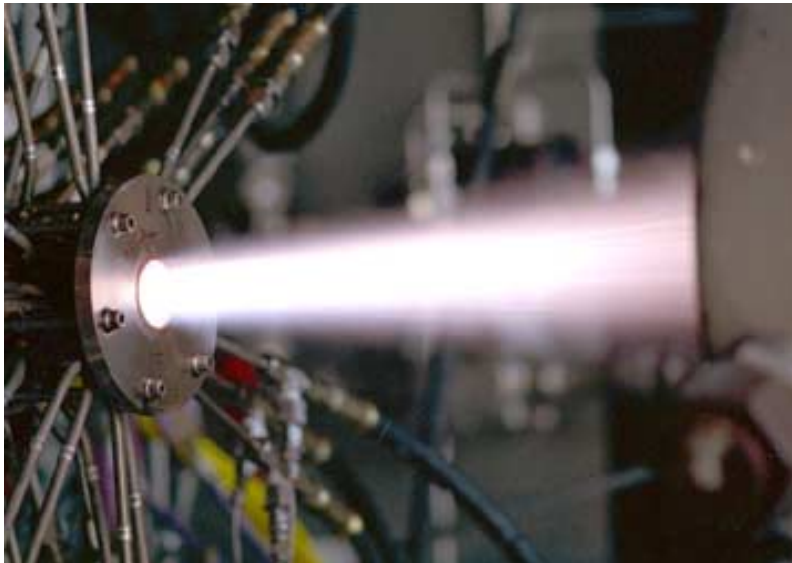
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- **Goal**
 - **Determine combustion and heat transfer characteristics of metallized gelled RP-1 /Al propellants in a rocket engine**
 - **Evaluate fuels including traditional RP-1 and metallized gelled RP-1 /Al with 0-, 5-, and 55-wt % loadings of aluminum, with gaseous oxygen as the oxidizer**
- **Hardware**
 - **Experiments conducted with a 40-lbf thrust engine composed of a modular injector, igniter, chamber, and nozzle**
 - **31 cooling channels for chamber calorimeter measurements, with temperature and pressure sensors**
- **Results**
 - **Gelled fuel coating, composed of unburned gelled fuel and partially combusted RP-1, formed in the 0-, 5- and 55-wt % engines**
 - **Coating caused a decrease in calorimeter engine heat flux in the last half of the chamber for 0- and 5-wt % RP-1 /Al propellants**

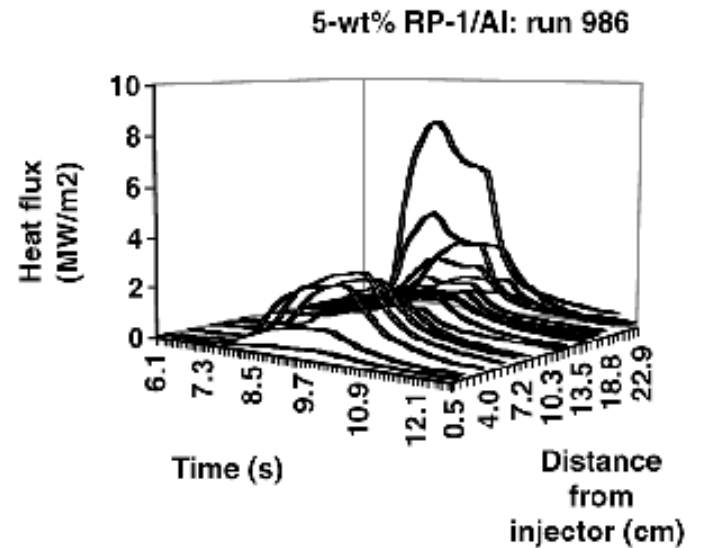


Metallized Gelled Propellants

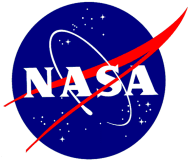
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5 wt% RP1-Al rocket engine test firing at the GRC.



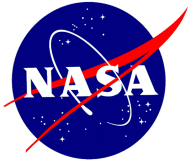
Three-dimensional roller-coaster plot of metallized gelled propellant heat flux: 5-wt % RP-1/Al.



The Benefits of Gelled and Metallized Gelled Propellants

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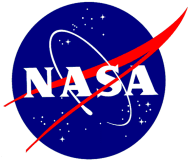
- **Gelled propellants**
 - Increased safety
 - Increased fuel density
 - Reduced leakage
 - Reduced slosh
 - Reduced cryogenic boiloff
 - Increases in engine specific impulse (in some cases)
- **Metallized Gelled Propellants**
 - All of the above and
 - Large increases in fuel density
 - Large increases in engine specific impulse (in some cases)



Propellant Technologies: Nanotechnology

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- **Nanotechnology is applied to rocket and aerospace propellants**
 - **Gellants**
 - **for hydrocarbons**
 - **for liquid cryogenic fuels (hydrogen, methane, propane)**
 - **Nanoparticulate materials used for gellants (hydrocarbon alkoxides)**
 - **Gellants increase fuel safety, density, and energy**
 - **Gellants reduce fuel slosh, and reduce the vehicle dry mass with higher fuel density**
 - **Nanophase aluminum particles added to rocket and aerospace fuels**
 - **Metallized gelled propellants**
 - **RP-1 /Aluminum**
 - **Hydrogen / Aluminum**
 - **Others (MMH / Aluminum, etc.)**
 - **Adding metal particles can increase engine exhaust velocity and fuel density**



The Benefits of Gelled and Metallized Gelled Propellants

NASA John H. Glenn Research Center, Lewis Field
Turbomachinery and Propulsion Systems Division

- **Gelled Propellants**
 - **Increased safety**
 - **Increased fuel density**
 - **Reduced leakage**
 - **Reduced slosh**
 - **Reduced cryogenic boiloff**
 - **Reduction in engine specific fuel consumption (in some cases)**
 - **Increases in engine specific impulse (in some cases)**
- **Metallized Gelled Propellants**
 - **All of the above and**
 - **Larger increases in fuel density**
 - **Larger reductions in engine specific fuel consumption (in some cases)**
 - **Larger increases in engine specific impulse (in some cases)**